Studies on soluble ribonucleic acid of rabbit liver. Action of polynucleotide phosphorylase

It has recently been established that specific amino-acid-activating enzymes catalyze both the formation of amino acid adenylates and the transfer of the amino-acyl moiety to RNA to form an amino acid-RNA compound^{1,2}. Only cytoplasmic or soluble RNA appears to function as an acceptor in this way and there is evidence that there exist different s-RNAs for different amino acids^{2–5}. The amino acid presumably is bound to the polynucleotide chain through the unesterified 2'- or 3'-hydroxyl group of the terminal nucleotide (nucleoside end)^{3,6,7}.

Both venom phosphodiesterase⁸ and polynucleotide phosphorylase⁹ act by stepwise removal of mononucleotide units starting from the nucleoside end of a polynucleotide chain. Grunberg-Manago observed that s-RNAs from different sources are slowly and incompletely phosphorolyzed by polynucleotide phosphorylase whereas other RNAs are degraded to completion¹⁰. Preiss et al.³ reported that s-RNA from Escherichia coli loses its ability to accept amino acids after minimal degradation (less than 5 %) with venom diesterase. We have confirmed this observation with rabbit liver s-RNA. It was of interest, therefore, to investigate more extensively the action of polynucleotide phosphorylase on s-RNA, particularly with respect to its effect on amino acid acceptor ability.

As shown in Table I the phosphorolysis of rabbit liver s-RNA* stops at 20–30 % of completion. The cessation of the reaction is not caused by enzyme inactivation or inhibition since (I) addition of fresh enzyme after 5 h produces no additional breakdown; (2) polyadenylic acid¹¹ added after 5 h is rapidly phosphorolyzed; and (3) the acid-soluble reaction products are not inhibitory. The extent of phosphorolysis

TABLE I PHOSPHOROLYSIS OF s-RNA

Expt.	Time h	μmole ³² P _i incorporated	% phosphorolyzed
	o	0.00	
A	5	0.12	28
	6	0.12	28
В	3	0.09	16
	6	0.11	20
	22	0.10	18

The reaction mixtures (0.20 ml) contained 0.1 M tris(hydroxymethyl)aminomethane buffer, pH 8.2, 10 mM MgCl₂, 0.2 mM ethylenediaminetetraacetate and potassium [\$^{32}P]phosphate buffer, pH 7.4 (23 mM in Expt. A and 10 mM in Expt. B). The amounts of s-RNA, in μ mole organic P, were 0.43 and 0.56 in Expts. A and B respectively. 90 and 210 μ g of a preparation of A. agilis polynucleotide phosphorylase of specific activity 23 (ref. 11) were used in Expts. A and B respectively. After incubation at 37°, 0.05 ml aliquots were added to 1.00 ml cold 2.5% HClO₄ and the precipitate was removed by centrifugation. The supernatant fluids were treated with charcoal to determine 32 P in acid-soluble nucleotides 12 . The results are given as μ mole 32 P₁ incorporated into acid-soluble nucleotides per 0.2 ml reaction mixture.

Abbreviations: RNA, ribonucleic acid; s-RNA, RNA from the soluble portion of a tissue homogenate; P_1 , inorganic orthophosphate.

^{*} The s-RNA used in these experiments, prepared from rabbit liver by NaCl extraction, contained less than 0.01 μ mole of bound amino acids per 25 μ moles of s-RNA ribose.

is not dependent on the concentration of s-RNA and is unaffected by mild alkaline treatment of s-RNA.

The portion of the s-RNA that remains undigested after treatment with polynucleotide phosphorylase retains its ability to accept activated amino acids. The specific activity* of s-RNA, with respect to proline or leucine, was exactly the same before and after phosphorolysis, namely, 914 \pm 58 and 906 \pm 48 for proline, 140 \pm 20 and 128 ± 24 for leucine; with respect to valine, on the other hand, the specific activity showed a small decrease, 96 ± 6 and 76 ± 8 .

It is apparent, therefore, that although venom diesterase and polynucleotide phosphorylase are thought to act in analogous fashions, the results of their action on s-RNA are quite different. The most likely interpretation of these findings suggests that polynucleotide phosphorylase degrades only a fraction of the polynucleotide chains present in the preparation and these to completion. 70-80 % of the chains may be entirely resistant to attack and the determining factor could be the extent of secondary structure (hydrogen bonding) in the resistant polynucleotide chains. The experiments of GRUNBERG-MANAGO¹⁰ and OCHOA¹³ have shown that multistranded polyribonucleotide chains are resistant to phosphorolysis. This hypothesis is supported by the observation that treatment of s-RNA with polynucleotide phosphorylase has essentially no effect on either the modal or the weight-average sedimentation coefficient (Table II).

TABLE II effect of polynucleotide-phosphorylase treatment on sedimentation of $s\text{-RNA}^{\star}$

	Sedimentation coefficient	
	W eight average	Mode
s-RNA (untreated)	4.0	4.0
s-RNA (undigested)	3.9	4.1

^{*} The absorbancy of the s-RNA solution was 1.0.

Measurements at 59,780 rev./min were made in 0.2 M NaCl, using the Spinco Model E Ultracentrifuge with u.v. optics. Values were computed from tracings obtained from the photographic images using the Spinco Model RA Analytrol, equipped with a microanalyzer attachment¹⁴.

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⁴ R. W. Holley and S. H. Merrill, J. Am. Chem. Soc., 81 (1959) 753.

¹ M. B. Hoagland, M. L. Stephenson, J. F. Scott, L. I. Hecht and P. C. Zamecnik, J. Biol. Chem., 231 (1958) 241.

P. Berg and E. J. Ofengand, Proc. Natl. Acad. Sci. U.S., 44 (1958) 78.
 J. B. Preiss, P. Berg, E. J. Ofengand, F. H. Bergmann and M. Dieckmann, Proc. Natl. Acad. Sci. U.S., 45 (1959) 319.

⁵ K. C. SMITH, E. CORDES AND R. S. SCHWEET, Biochim. Biophys. Acta, 33 (1959) 286.

⁶ H. G. ZACHAU, G. ACS AND F. LIPMANN, Proc. Natl. Acad. Sci. U.S., 44 (1958) 885.

^{*} Specific activity is expressed as counts/min/absorbancy in a hot-HClO₄ digest of s-RNA.

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    L. I. HECHT, M. L. STEPHENSON AND P. C. ZAMECNIK, Proc. Natl. Acad. Sci. U.S., 45 (1959) 505.
    W. E. RAZZELL AND H. G. KHORANA, J. Biol. Chem., 234 (1959) 2114.
    R. J. HILMOE, Ann. N.Y. Acad. Sci., 81 (1959) 660.
    M. GRUNBERG-MANAGO, J. Mol. Biol., 1 (1960) 240.
    M. GRUNBERG-MANAGO, P. J. ORTIZ AND S. OCHOA, Biochim. Biophys. Acta, 20 (1956) 269.
    M. F. SINGER, J. Biol. Chem., 232 (1958) 211.
    S. OCHOA, Arch. Biochem. Biophys., 69 (1957) 119.
    V. N. SCHUMAKER AND H. K. SCHACHMAN, Biochim. Biophys. Acta, 23 (1957) 628.
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Peptides isolated from peptic hydrolysate of diisopropylphosphoryl-trypsin

In the comparative studies on protein structures¹ which our laboratory has been working on we have been concerned above all with chymotrypsinogen and trypsin. A survey of the peptide sequences in these proteins so far known was published by us earlier².

The present paper gives a preliminary report on results obtained by studying peptides isolated from a peptic hydrolysate of DIP-trypsin. This study permits us to ascertain mutual bonds in the vicinity of the residues of lysine and arginine, which are split on tryptic hydrolysis.

DIP-trypsin (1000 mg) was hydrolyzed by pepsin (enzyme to substrate ratio, 1:50) at 37° and pH 2.0 for 2.5 h. A small quantity of non-dialyzing material was removed from the peptic hydrolysate by dialysis against distilled water. The freezedried dialysate was fractionated on a Zerolit 225 X2 ion-exchange resin (100–200 mesh) column of 2.2 × 150 cm at 40° using an elution gradient. For this purpose volatile buffers composed of pyridine and formic acid or pyridine and acetic acid proved satisfactory as in the case of the separation of a peptic hydrolysate of chymotrypsinogen³.

Elution was initiated with the o.I M buffer (with respect to pyridine) and the gradient developed by gradually introducing solutions of increasing molarity and pH-value into a mixing vessel (volume, 2200 ml). The solutions used were as follows: 0.2 M, pH 3.I; I M, pH 5.I; 2 M, pH 5.2; 2 M, pH 7.0 and 2 M ammonia. The eluates were collected in 20-ml fractions at a flow rate of I ml/min.

The course of the separation on the column was checked by submitting dry residues of 0.4-ml aliquots taken from each fraction to descending paper chromatography³ in the system n-butanol-pyridine-acetic acid-water $(15:10:3:12)^4$. According to the chromatograms, individual fractions obtained from the column were pooled and fractionated further by preparative chromatography on Whatman No. 1 or No. 3 paper in the system⁴ mentioned above. The zones obtained were purified by paper electrophoresis in the system formic acid-acetic acid of pH 1.9 at a potential gradient of 65 V/cm.

A majority of the zones isolated comprised individual peptides as was proved by calculating the molar ratios of amino acids by comparing the peptide hydrolysate

Abbreviation: DIP-, diisopropylphosphoryl-.